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Evaluation of brick kiln operation impact on soil microbial biomass and enzyme activity

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Abstract

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Keywords

Brick kiln cluster Agricultural soil Cd and Pb contamination Potential ecological risk index Microbial metabolic quotient ED₅₀ Heavy metal emission from brick kiln operation in developing countries is one of the major sources of environmental pollution. The present study evaluated the intensity of Cd and Pb pollution and the impact on soil microbial activity in agricultural soils in the vicinity of the brick kiln cluster of Hathazari, Chattogram, Bangladesh. It is a major concern as anthropogenic stress on soil microorganisms is directly related to crop productivity. Soil samples were collected from 21 sites covering 7 locations including the reference sites for the assessment of the toxic impact on soil biota. Soil samples were analyzed using standard procedures. In some of the sampling sites, Cd and Pb concentrations were significantly higher than the reference sites. Metal concentration indicates that the anthropogenic input in the soils was in the range of 0.27 to 1.07 mg·kg⁻¹ of Cd and 19.07 to 52.07 mg·kg⁻¹ of Pb. However, the concentrations of Pb were not in toxic concentration when compared to the standard level by Chinese environmental quality standards for soil. The highest contamination degree (PER) of the soils was 200.87 and the lowest was 115.83. The contamination factor demonstrated that the soils were in the moderate to considerable level of contamination. The results showed that the number of soil microbial population, microbial activity, microbial biomass carbon, dehydrogenase, urease, acid phosphatase and arylsulfatase activities in the reference soil were all higher than in the agricultural soil in the vicinity of brick kiln cluster. Exponential curves showed a significant positive correlation between heavy metal and microbial metabolic quotient (qCO₂) indicating metal stress and high concentration of heavy metals decreased microbial biomass and enzyme activity. Soil pH and Cd content were identified as the key influential factors controlling soil biological functions. A significantly high correlation was observed for Cd and Pb (r = 0.89, p < 0.001), it suggests the same source of contamination input. Contamination of Cd and Pb is attributed to heavy input of aerial deposits of metal-enriched fumes from brick kiln operation. A significant negative impact of Cd and Pb on soil microbial activities and enzyme activities was also profound from correlation studies and PCA analysis. However, regular application of fertilizer in agricultural soils may have supported adaptation to long-term Cd stress mainly through the maintenance of microbial activity. The study is important in eco-toxicological and biomonitoring aspects as the data on heavy metal toxicity to the soil environment can act as guidelines for the continuation of brick kiln operation and the sustainable utilization of natural resources.

1. Introduction

Environmental pollution is a global problem. However, the source and increase in pollution in developing countries are different and drastic compared to the developed countries. The environmental impact of brick production has become an earnest concern. Air pollution and land degradation from brick kiln's emission are regular in developing countries, adversely affecting the environment of its surroundings (Biswas et al., 2018). Rapid urbanization created a booming in the construction industry. Brick is the principal building material for some countries where stone aggregate is not available. The brick sector in a developing country is specified by firing technologies, environmental pollution, reliance on manual labor and use of soil as essential raw material. Brick kilns are very energy-intensive and highly polluting, they usually burn lowquality coal with high sulfur (about 5 percent), clinker, wood dust, furnace oil, rubber tires, plastics and wood (Tusher et al., 2018). The incomplete combustion process of brick kiln furnaces releases solid particles and greenhouse gases. The pollutants are suspended particulate matter, carbon monoxide, carbon dioxide and heavy metals in fumes (Skinder et al., 2014). The large brick kiln clusters, located in the vicinity of large brick demand centers are a major cause of concern, the pollution impact from the emission from single isolated brick kilns located in rural areas is not significant.

Long term brick kiln operation can cover the surface area of the local vegetation and soil with brick kiln dust which may cause

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significant change to the physicochemical properties as well as the nutrient status of nearby soil with ultimate rigorous impact on soil ecology (Islam et al., 2015; Sarkar et al., 2016). The burning of enormous carbon (C) and nitrogen (N) in brick kiln degrades the soils and led to a threat for atmospheric pollution and climate change (Khan et al., 2007). Which further results in crop yield reduction, marginal vegetation growth as well as soil degradation and reduction of microbial activity and nutrient cycling (Sharma, 2000). Mobility and bioavailability of heavy metals are considerably increased in comparison with the original waste by the incineration process. Heavy metals are generated from the brick furnace as aerial pollutants deposited shortly to the surroundings and distributed in soils and water sources. An enormous amount of heavy metal can be deposited in the biosphere as the loading rate is approximately 20 times higher than the removal rate (Ravankhah et al., 2017). Heavy metals as a soil pollutant can be an issue of concern for their persistence, toxicity and bioaccumulation (Li et al., 2015). The concentration of Cd and Pb in agricultural soil of brick kiln area was outlined to be more than the regulatory standards imposed by the US Environmental Protection Agency (Ismail et al., 2012).

Bangladesh is the fourth-largest producer of clay fired bricks in the world after China, India and Pakistan. There are more than 7,000 brick kilns, producing about 27 billion bricks annually (Eil et al., 2020). Brick kiln emission is the largest source of greenhouse gas emissions in Bangladesh estimated to be 15.67 million tons of CO_2 annually (Imran et al., 2014). Traditional kilns in Bangladesh have particulate emissions above 1,000 mg per m³ and coal consumption of 20–22 tons per 100,000 bricks produced (Haque et al., 2018). A plethora of information is available for heavy metal pollution in agricultural soils of Bangladesh (e.g. Chowdhury and Rasid, 2016; Ahmed et al., 2018; Proshad et al., 2019; Mallick et al.,

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2019; Hasan et al., 2020; Alam et al., 2020). Some studies focused on heavy metal contamination of agricultural soils by heavy traffic (e.g. Aktaruzzaman et al., 2013; Zakir et al., 2014; Tasrina et al., 2015). Elevated concentrations of heavy metals were detected in some of the studies, which have contributed to the declined soil health and crop yields. However, no detailed study on soil microbial biomass and enzyme activity in agricultural soil concerning heavy metals impact from brick kiln clusters have been conducted yet. Several huge clusters of the brick kiln are present in Hathazari, Chattogram which can be an important source of Cd and Pb toxicity with the potential to harm the ecology of the surrounding agricultural soil (Ravankhah et al., 2017). Soil degradation and environmental pollution are not inevitable but can be controlled if major abuses are avoided and improved methods of environmental management are developed. With this background, the stress effect of Cd and Pb on soil microbial and enzyme activities was evaluated.

2. Materials and methods

2.1. Study area and soil sampling

A cluster of the brick kiln was selected which were surrounded by agricultural fields in the Charia area, Hathazari, Chittagong district, Bangladesh. The sampling from the study area was carried out during the dry seasons as the brick kilns are the major source of air pollution during the manufacturing season of October to March, depending on the monsoonal rains. Emitterinfluenced sampling locations were set based on crops growing and distance from the brick kiln cluster (A-F) (Table 1). At these locations, three different crop fields were selected and three soil

Table 1

Physicochemical properties (mean ± SD) of different agricultural soils adjacent to the brick kiln cluster, Hathazari, Chattogram

Each mean is the average of the values obtained for three samples of each soil. Values in the same column followed by the same letter(s) are not significantly different at p < 0.05 according to ANOVA.

[OM = Organic matter, Total N = Total nitrogen, Av P = Available phosphorus, Total Pb = Total lead concentration, Total Cd = Total cadmium concentration, SD = Standard deviation]

Location	Sites	рН	Clay %	OM %	Total N %	AvP mg∙kg⁻¹	Total Pb mg∙kg⁻¹	Total Cd mg∙kg⁻¹
	S1	$4.87{\pm}0.08^{\rm f}$	22.82 ^{ab}	$0.67{\pm}0.02^{ab}$	$0.16 \pm 0.00^{\mathrm{g}}$	7.14 ± 0.10^{g}	45.39±0.05°	$1.00\pm0.01^{\text{m}}$
A	S2	5.39 ± 0.03^{i}	27.71^{cde}	$1.19{\pm}0.05^{\rm i}$	$0.19{\pm}0.01^{ij}$	8.92 ± 0.79^{i}	38.83 ± 0.02^{n}	$0.63{\pm}0.06^{\rm h}$
	S3	5.26 ± 0.03^{h}	23.92 ^{abc}	0.88 ± 0.03^{d}	$0.12{\pm}0.01^{\rm de}$	$6.87{\pm}0.11^{\rm fg}$	$27.33{\pm}0.03^{\rm hi}$	$0.98 \pm 0.01^{\text{m}}$
	S4	4.15 ± 0.05^{a}	24.25 ^{abc}	$1.11{\pm}0.05^{\rm gh}$	$0.15{\pm}0.00^{\rm f}$	$3.64 \pm 0.34^{\text{b}}$	28.38±0.65 ^{ij}	$0.66{\pm}0.02^{\rm h}$
В	S5	$5.62{\pm}0.00^{\rm k}$	30.67^{defg}	$1.44{\pm}0.00^{\rm k}$	$0.19{\pm}0.00^{\rm i}$	12.18 ± 0.21^{k}	$25.34 \pm 0.33^{\rm f}$	$0.46{\pm}0.01^{\rm f}$
	S6	5.07 ± 0.02^{g}	22.84 ^{ab}	$0.95{\pm}0.05^{\rm de}$	$0.11{\pm}0.01^{\rm ab}$	9.48 ± 0.06^{j}	52.07 ± 0.03^{p}	1.07 ± 0.01^n
	S7	4.44 ± 0.11^{b}	34.13 ^g	0.76±0.00°	$0.13 \pm 0.00^{\text{e}}$	2.79 ± 0.38^{a}	$31.40{\pm}0.07^{\rm k}$	$0.71{\pm}0.00^{\rm i}$
С	S8	5.76 ± 0.08^{1}	27.33^{bcd}	1.22 ± 0.06^{i}	$0.19{\pm}0.00^{\rm i}$	$6.45{\pm}0.02^{\rm f}$	$26.17 \pm 1.27^{\text{fg}}$	$0.55 \pm 0.00^{\text{g}}$
	S9	$6.30\pm0.02^{\circ}$	32.83 ^g	1.67 ± 0.07^l	0.20 ± 0.00^{j}	$13.44{\pm}0.07^{\rm l}$	20.75 ± 0.14^{d}	$0.31{\pm}0.05^{\rm cd}$
	S10	$4.56{\pm}0.03^{\rm cd}$	32.35^{fg}	0.91 ± 0.07^{d}	$0.11{\pm}0.01^{\rm bc}$	2.65 ± 0.38^{a}	33.52 ± 0.10^{1}	$0.88{\pm}0.01^{\rm k}$
D	S11	5.07 ± 0.09^{g}	22.82 ^{ab}	$1.06{\pm}0.00^{\rm fg}$	$0.16{\pm}0.01^{\text{gh}}$	$4.38{\pm}0.16^{\rm cd}$	22.05 ± 0.02^{e}	$0.31{\pm}0.01^{\rm d}$
	S12	$5.37{\pm}0.02^{\rm i}$	27.74^{cdef}	1.36 ± 0.08^{j}	$0.23{\pm}0.01^{\rm l}$	5.43 ± 0.08^{e}	20.73 ± 0.04^{d}	$0.28{\pm}0.01^{\rm cd}$
	S13	$4.77{\pm}0.06^{\rm e}$	22.95 ^{ab}	$1.01{\pm}0.05^{\rm ef}$	$0.15{\pm}0.00^{\rm f}$	5.70 ± 0.10^{e}	29.16 ± 0.03^{j}	$0.70{\pm}0.01^{\rm i}$
E	S14	$4.55{\pm}0.04^{\rm cd}$	22.63ª	$0.65{\pm}0.05^{\rm ab}$	$0.11{\pm}0.01^{\rm bc}$	3.83 ± 0.26^{b}	37.54 ± 0.38^{m}	0.92 ± 0.01^{1}
	S15	$5.50{\pm}0.01^{\rm j}$	27.02^{abcd}	$1.22{\pm}0.00^{\rm i}$	$0.17{\pm}0.00^{\rm h}$	$3.89{\pm}0.28^{\rm bc}$	19.07±0.02°	$0.27\pm0.02^{\circ}$
	S16	$4.58{\pm}0.10^{\rm d}$	22.63ª	$0.60 \pm 0.00^{\mathrm{a}}$	$0.11{\pm}0.00^{\rm abc}$	4.56 ± 0.13^{d}	32.55 ± 0.53^{1}	$0.82{\pm}0.01^{\rm j}$
F	S17	$5.89 \pm 0.04^{\text{m}}$	22.63ª	0.76±0.00°	$0.12{\pm}0.01^{\text{cd}}$	$8.05{\pm}0.51^{\rm h}$	22.07 ± 0.03^{e}	0.37 ± 0.02^{e}
	S18	$4.48{\pm}0.02^{\rm bc}$	22.51ª	$0.68{\pm}0.00^{\rm b}$	0.10 ± 0.01^{a}	7.92 ± 0.36^{h}	$27.19{\pm}1.42^{\text{gh}}$	$0.56 \pm 0.03^{\text{g}}$
	S19	$5.56{\pm}0.02^{\rm jk}$	27.90^{cdef}	$1.16{\pm}0.00^{\rm hi}$	$0.21{\pm}0.00^{\rm k}$	14.35 ± 0.49^{m}	$9.22\pm0.06^{\mathrm{b}}$	$0.17{\pm}0.01^{\rm b}$
R	S20	6.07 ± 0.02^n	26.20^{abcd}	$1.22{\pm}0.06^{\rm i}$	0.22 ± 0.00^{1}	15.12 ± 0.04^{n}	9.01 ± 0.77^{b}	$0.13{\pm}0.00^{\rm a}$
	S21	5.88 ± 0.01^{m}	32.03^{efg}	1.22 ± 0.06^{i}	0.22 ± 0.01^{1}	$14.48{\pm}0.07^{\rm m}$	5.96 ± 2.07^{a}	$0.14{\pm}0.00^{\rm ab}$

subsamples were collected from each site. The bright yellow spots on the (Fig. 1) are the sampling sites. The reference location (R) was the agricultural soils approximately 2.5 km far from location A. Wind flow direction was considered for the effect of fly ash deposition on the soil. Soil samples were collected from the soil surface to the root zone (0–30 cm). Soil sampling was done with a stainless-steel spade. All of the samples were put in polythene bags and transported to the laboratory on the day of sampling.

2.2. Processing of samples

The composite soil samples were sieved through a 2 mm sieve, homogenized, a portion of this was air-dried for physical and chemical analysis, another portion was ground to pass through a 1 mm sieve, adjusting to 45% of water holding capacity, stored in polythene bags at 4°C before soil microbial and enzymatic activities analysis.



Fig. 1. Location of the study area Hathazari, Chittagong, Bangladesh. Brick kilns can be prominently seen in the Satellite image. Map of Bangladesh also showing the wind flow as a climatic element of Bangladesh (website 1). Balloons denote locations and pins denote sites

2.3. Measurement of soil physical and chemical properties

General soil characteristics were determined following the standard procedures. The pH of the soil samples was measured by pH meter at dry soil and distilled water ratio of 1:5 as described in Jackson (1973). Particle size distributions of the soils were determined by the hydrometer method (Day, 1965). Textural classes were determined using 'soil automatic texture calculator' by Natural Resources Conservation Service Soils of the United States Department of Agriculture (website 2). The organic carbon (OC) content of the soil samples was determined volumetrically by the wet oxidation method by Nelson and Sommers (1982). Organic matter (OM) content was estimated through the use of an approximate correction factor, the 'Van Bemmelen factor' of 1.724 which is based on the assumption that OM contains 58% OC. Total nitrogen (TN) content in soil was determined by the Micro-Kjeldahl method following H₂SO₄ acid digestion and alkali distillation and available phosphorus (AvP) by the colorimetric method after digestion with hydrofluoric and perchloric acid (Jackson, 1973). The total concentrations of cadmium (Cd) and lead (Pb) were determined by Atomic Absorption Spectrophotometer (Aligent 240) after strong acid digestion (1:1 mixture of concentrated nitric and perchloric acids) of 200 mg of soil samples. The digested samples were filtered and collected in 5 ml of 2.0 M HCL as in Ure (1990).

2.4. Measurement of soil microbiological properties

Numbers of total bacteria and fungi in soils were counted using the dilution plate method as described in Johnson and Curl (1972). Nutrient agar (NA) medium was used with bacteria and potato dextrose agar (PDA) medium with fungi. Three plates were used for each soil. The plates were incubated at 28°C for 7-10 days and counting made for forming colonies. Soil microbial biomass (MBC) was measured by the method described by Anderson and Ingram (1993). The microbial cells in soil were killed by fumigation with ethanol-free chloroform. Immediately after pre-incubation, duplicate portions soil, 5 g for each were taken in falcon tubes. One set of samples was fumigated with ethanol-free chloroform for 24 h at 25°C in a sealed desiccator. Non fumigated set of samples in falcon tubes were capped and stored at 8°C. After fumigant removal, both fumigated and non-fumigated soils were extracted with freshly prepared 0.5 M potassium sulfate at 1:4 ratios and filtered. Dissolved OC in the extracts was determined after dichromate digestion by titrating with 0.03 M acidified ferrous ammonium sulfate. The amount of soil MBC was calculated from the difference between the extracted carbon from chloroform fumigated and non-fumigated samples. Soil microbial activity (MA) was determined by trapping the CO₂ in NaOH which were evolved from the soil during incubation in a closed system (Alef, 1995). The trapped CO₂ was determined by measuring electrical conductivity (Rodella and Saboya, 1999). For this purpose, 50 g (oven-dry basis) moist preincubated (60% of water holding capacity for 10 days) soil was placed in 1-liter capacity incubation Jars. Ten ml of 1.0 M NaOH solution in 50 ml falcon tubes were placed in each jar as the CO₂ trap. A falcon tube with water was added into the jar to maintain the soil moisture. Jars were made airtight immediately. Two jars with 1.0 M NaOH but without soil were used as controls. All jars were incubated at 25°C. CO_2 absorbed in traps were analyzed at 1, 7, 14, 30 days of NaOH placement. Each time fresh NaOH solution (10 ml) was replaced to trap CO_2 for the next days. In this method, CO_2 evolved from each sample was calculated as the difference between the initial and the CO_2 concentration after each measurement period. The substrate induced respiration (SIR) of the soils was assessed according to the rate of the maximal initial respiration of the microorganism after the enrichment of the soils with 0.5% glucose (West and Sparling, 1986). Over the first 2 h, the increase in CO_2 –C is proportional to the size of the initial MBC concentration. Respiration was determined by trapping the CO_2 in NaOH as in MA.

2.5. Measurement of soil enzymatic properties

Soil dehydrogenase activity was determined by the procedure of Casida (1977). Soil samples were suspended in a triphenyl tetrazolium chloride solution and incubated for 6 h at 37°C. The triphenyl formazan (TPF) produced was extracted with methanol and measured photometrically at 485 nm. Urease activity was assayed according to the method of Tabatabai and Bremner (1972). After the addition of a buffered urea solution, soil samples were incubated for 2 h at 37°C. The filtrated solution was distilled with MgO. The produced NH₄+-N was collected into a boric acid indicator solution and titrated with diluted H_2SO_4 to determine the NH_4^+ -N. Acid phosphatases activity was measured using the method of Eivazi and Tabatabai (1977). After the addition of a buffered p-nitrophenyl phosphate solution (pH 6.5), soil samples were incubated for 1 h at 37°C. The p-nitrophenol released by phosphomonoesterase activity was extracted and colored with NaOH and was measured photometrically at 400 nm. Arylsulfatase activity was measured by the potassium p-nitrophenyl sulfate method (Tabatabai and Bremner, 1970). After the addition of a buffered potassium p-nitrophenyl sulfate (pH 5.8), soil samples were incubated for 1 h at 37°C. The P-nitrophenol released by phosphomonoesterase activity was extracted and colored with NaOH and was measured photometrically at 400 nm.

2.6. Ecological risk assessment for soil pollution

The degrees of heavy metal contamination in agricultural soils can be evaluated with a comprehensive potential ecological risk (*PER*) index. The sensitivity of the biological community can be expressed by it to the heavy metal stress and indicates the potential ecological risk caused by the overall heavy metal contamination. The equations which were used to calculate *PER* are as follows (Guo et al., 2010):

$$C_{f}^{i} = \frac{C^{i}}{C_{n}^{i}} \qquad C_{d} = \sum_{i=1}^{n} C_{f}^{i} \qquad E_{r}^{i} = T_{r}^{i} \times C_{f}^{i} \quad PER\sum_{i=1}^{m} E_{r}^{i}$$
(1)

where: *PER* = comprehensive potential ecological risk index, C_f^i = single heavy metal contamination factor, C^i = content of the heavy metal in samples, C_n^i = background value of the heavy metal, C_d = degree of contaminations, E_r^i = potential ecological risk index, T_r^i = biological toxic factor, the biological factors for cadmium = 30 and lead = 5 (Guo et al., 2010; Islam and Hoque 2014).

2.7. Microbial metabolic quotient

Stress in the microbial population can be determined by the microbial quotient (qMic) and metabolic quotient (qCO₂). Organic carbon in soil generally undergo microbial synthesis and converted to humus. But, in the case of increased stress, more CO₂-carbon per unit microbial biomass per unit time is produced to counter stress. Metabolic quotient was calculated from basal respiration at the end of the 30 days' incubation period according to the following equation (Anderson and Domsch, 1990).

$$q \text{CO}_2 = \frac{r}{\text{MBC}}$$
(2)

where: r = respiration rate, mgC·g soil·24h⁻¹, MBC = soil microbial biomass carbon, mg biomass C·kg soil⁻¹.

Microbial quotient representing the ratio of soil MBC to organic carbon expressed as percent (%) to explore the percent of organic carbon present as microbial biomass carbon (Anderson and Domsch, 1989):

$$q\text{MIC} = \frac{\text{MBC}}{\text{OC}}$$
(3)

where: qMic = microbial quotient, MBC = soil microbial biomass carbon, mg biomass C·kg soil⁻¹, OC = total organic carbon, mg·kg⁻¹.

2.8. Kinetic models related to enzymatic activity

The inhibition of enzymatic activity by heavy metal was assessed by two kinetic models (Model 1: Equation 4 and Model 2: Equation 5) and a sigmoidal dose-response model (Model 3: Equation 6) using *PER* (Gao et al., 2010). The 50% ecological dose (ED₅₀) values are calculated for Models 1 and 2 by fitting Equation 7 and for Model 3 by fitting Equation 8:

Model 1:
$$v = \frac{c}{1 + bPER}$$
 (4)

Model 2:
$$v = \frac{c(1 + aPER)}{1 + bPER}$$
 (5)

Model 3:
$$v = \frac{x}{1 + e^{y(l-z)}}$$
 (6)

$$ED_{50} = \frac{1}{b}$$
(7)

$$ED_{50} = c^c \tag{8}$$

where, ED_{50} = total ecological toxicity coefficients which lead to enzyme activity inhibited by 50%, *PER* = potential ecological risk index under multiple heavy metal pollution in Equation 1, v = response variable, a, b and c = fitting parameters with positive values and b > a, l = natural logarithm of *PER*, x = uninhibited value of v, y = slope factor, z = natural logarithm of ED_{50} .

2.9. Statistical analysis

All the results are expressed on an oven-dry weight basis which was measured with three replications. Correlations between the selected parameters and standard deviation were determined using Microsoft Excel 2016 program. Regression between soil parameters and soil microbial activities and enzyme activities was fitted to linear and exponential functions. The effects of Cd and Pb were determined by one-way analysis of variance (ANOVA) and the significance of the parameters was tested using the least significant difference multiple range test at p < 0.05. Pared-samples T-test measured for soil samples firstly by considering soil samples all together (n = 63), secondly by considering mean values representing the sites with different agricultural management (n = 21) and thirdly considering specific location types (n = 7) by IBM SPSS program. The dendrogram grouping for cluster analysis was performed by IBM SPSS and Principal Component Analysis (PCA) was performed by XLSTAT.

3. Results and discussion

A detailed study on soil properties of a cluster of brick kiln in the Charia area, Hathazari, Chittagong district was done and its operation effect on soil properties and microbial biomass and activity as well as soil enzyme activity in the surrounding agricultural area was analyzed to infer the impact of brick kiln emissions.

3.1. Physical and chemical properties of the agricultural soils

The soil texture of the agricultural soil was sandy clay loam. This textural class of soil is favorable for agricultural practice in Bangladesh. The percentage of clay ranged from 22.51 to 34.13% (Table 1). There was no significant difference between reference soil (28.71%) and agricultural soils (26.10%) around the brick kiln cluster for average clay content. The pH values of the samples ranged from 4.15 to 6.30 (Table 1), which were very strongly acidic to neutral (Hazelton and Murphy, 2016). Reference soils (average pH 5.84) were moderately acidic in nature. SRDI (2008) reported that the pH values of the soil series (Lama, Rangamati, Matiranga) of the Hathazari area are strongly acidic in nature. The inherited acidic nature of some of the studied soils can be increased due to fly ash application to the agricultural field which is produced as a byproduct from brick kiln operation (Fatima et al., 2011). The pH of soil samples at agricultural soils varied significantly with each other as well as with the reference sites, so it refers to impact from the brick kiln.

Soil OM and TN ranged from 0.60 to 1.67% and 0.10 to 0.23%, respectively under agriculture soils near the brick kiln cluster (Table 1). Reference sites (S19, S20 and S21) and sites S5, S8, S9, S12 and S15 had the higher OM (\geq 1.20%) and TN (\geq 0.20%) contents. The content of OM at all the sites was nearly uniform, there were no significant differences between average OM content of agricultural soils and reference soils. The overall simi-

lar value of OM may be due to similar cultivation induced OM decomposition and excessive use of fertilizers in the cultivated field. However, some variations of TN in the same location may be the result of the variation of agricultural practices. Deposition of unburnt hydrocarbon in the form of soot on soil from aerial deposits from brick kilns and vehicles running in Chittagong-Bandarban highway can also be considered for some contribution of OC content in some sites (Bisht and Neupane, 2015). The AvP content also varied among the agricultural soils (Table 1). The AvP content of the samples was within the range of 2.65 to 13.44 mg·kg⁻¹ in the agricultural soils near brick kiln and of 14.65 mg·kg⁻¹ (average) in the reference soils. Soils of the agricultural sites (S4, S7, S10, S11, S14, S15 and S16) showed deficient in AvP content as a critical level of AvP is below 7.00 mg·kg-1 (BARC 2012). Regular application of P fertilizer to the agricultural field may be the reason for the irregular pattern of phosphorus to the agricultural sites. Soil pH was highest in site S9 (6.30) in combination with the highest OM (1.67%) and AvP (0.20%). Essential nutrient elements – TN (r = 0.79, p < 0.001) and AvP (r = 0.51, p < 0.001) were significantly correlated to soil OM content. TN and AvP were significantly correlated to each other (r = 0.60, p < 0.001) and a positive significant (r = 0.67, p < 0.001) correlation of OM and pH was found for the soils (Fig. 2a).

3.2. Cd and Pb in the agricultural soil

The results of Cd and Pb concentration in the soil samples are presented in (Table 1). The concentration of Cd in agricultural soils of the study area ranged from 0.27 to 1.07 mg·kg⁻¹ and 19.07 to 52.07 mg·kg⁻¹ for Pb. The average concentrations of Cd and Pb were 0.64 and 29.97 mg·kg⁻¹ respectively. In reference soils, the values were 0.15 mg·kg⁻¹ soil for Cd and 8.06 mg·kg⁻¹ soil for Pb. In particular, the mean concentrations of Cd and Pb in the study area were 4.3 and 3.7 fold higher than the refer-

50	IL	SCI	EN	CE	AN	NUAL

ence sites, respectively. Therefore, the anthropogenic activities related to brick kiln operation may be responsible for the Cd and Pb contamination in the agricultural soils. Heavy metal concentrations in the study region were also compared to the standard values for agricultural soil (Chinese environmental quality standards for soil, Pb is 250 mg·kg⁻¹ and Cd is 0.30 mg·kg⁻¹ for dry agricultural soil at pH < 6.5, Act no. 220/2004 coll. of laws) (Chen et al., 2018). Comparing the concentrations sites S6, S1, S3, S2, S14, S10, S16, S7, S13 and S4 seem to have a higher contamination of Cd and Pb that was attributed to the heavy input of aerial deposits of metal-enriched fumes. The results showed an irregular pattern of the heavy metal (Cd and Pb) availability in soil samples. The PER which shows the extent of contamination of the sampling sites indicated moderate to a considerable level of contamination in the agricultural soils around the brick kiln cluster. However, no potential ecological risk was found with reference sites (S19, S20 and S21), for which the sum of toxic units was lower than 40.

The pH values of the studied area were naturally acidic. Soil pH showed a significant negative correlation with Cd (r = -0.65) and Pb (r = -0.56), which may suggest that pH influenced the distributions of these metals in soils. For both heavy metals, Cd and Pb, an inversely proportional relation between their accumulation rate and concentrations of OM was found (Fig. 2a), which indicates that with an increase in Cd and Pb concentration, a significant decrease in the OM of soil was observed, which might be due to low soil biological activity. Furthermore, both Cd and Pb accumulation rates in soil decreased with the increasing concentrations of clay in that soil. The correlation between metal concentration in soil and clay content and soil OM depends on the physical and chemical characteristics of the soil, mostly soil pH and the content and type of clays, the content and type of organic matter and oxides (Orrońo and Lavado, 2009). Rieuwerts et al., (1998) reported that the correlation of heavy metals was found

	Clay	BD	WHC	Ηd	MO	NT	AvP	Soil Pb	Soil Cd		Bacteria	Fungi	SIR	МА	MBC	Dehydro	Urease	AcidP	ArylS
Clay	1.0	-0.5	0.5	0.3	0.5	0.4	0.2	-0.3	-0.3	Bacteria	1.0	0.7	0.6	0.5	0.8	0.7	0.3	0.6	0.6
BD	-0.5	1.0	-0.6	-0.3	-0.5	-0.5	-0.4	0.5	0.4	Fungi	0.7	1.0	0.6	0.4	0.7	0.6	0.2	0.5	0.5
WHC	0.5	-0.6	1.0	0.5	0.6	0.6	0.4	-0.4	-0.4	SIR	0.6	0.6	1.0	0.2	0.7	0.3	0.3	0.5	0.4
pН	0.3	-0.3	0.5	1.0	0.7	0.7	0.7	-0.6	-0.7	МА	0.5	0.4	0.2	1.0	0.6	0.7	0.7	0.2	0.4
ОМ	0.5	-0.5	0.6	0.7	1.0	0.8	0.5	-0.5	-0.6	MBC	0.8	0.7	0.7	0.6	1.0	0.8	0.5	0.7	0.7
TN	0.4	-0.5	0.6	0.7	0.8	1.0	0.6	-0.6	-0.7	Dehydro	0.7	0.6	0.3	0.7	0.8	1.0	0.5	0.5	0.6
AvP	0.2	-0.4	0.4	0.7	0.5	0.6	1.0	-0.5	-0.5	Urease	0.3	0.2	0.3	0.7	0.5	0.5	1.0	0.1	0.2
Soil Pb	-0.3	0.5	-0.4	-0.6	-0.5	-0.6	-0.5	1.0	0.9	AcidP	0.6	0.5	0.5	0.2	0.7	0.5	0.1	1.0	0.5
Soil Cd	-0.3	0.4	-0.4	-0.7	-0.6	-0.7	-0.5	0.9	1.0	ArylS	0.6	0.5	0.4	0.4	0.7	0.6	0.2	0.5	1.0

Fig. 2. Pearson's correlation analysis correlating (a) soil physicochemical properties and heavy metals (b) Soil microbiological properties and enzyme activities (n = 63) (p < 0.05). Green boxes show positive correlations; red boxes show negative correlations

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to be high with Fe–Mn oxides than with SOM in some soils. The influence of clay on Cd and Pb accumulation can be negligible or negative (Wieczorek et al., 2018) and metal sorption is more related to clay mineralogy than clay quantity (Orrońo and Lavado, 2009). Accumulation and activity of anthropogenic Cd and Pb in the soil also influence by the size and surface area of the emitted particles, differences in speciation and type of binding forms (Rieuwerts et al., 1998). In the case of soil samples, Pb showed strong significant correlations with Cd (r = 0.89) indicating their associations among themselves and hence, may be originated from a similar source of contamination.

3.3. Microbiological properties of the agricultural soils

Different heavy metal stresses which may enter into the soil system by anthropogenic activities can affect soil microorganisms, consequently, negatively affect soil fertility via reduced OM decomposition and nutrient (Sharma et al., 2017). Microbial activities found to be decreased in the highly contaminated agricultural soils and significant negative relations were found between soil Cd and Pb contents and microbial numbers and activities (Table 3). The mechanism involved in inactivating and inhibiting soil microbial activity differs for different heavy metals. The number of culturable bacteria ranged from 61×10^5 CFU·g⁻¹ dry soil to 302×10^5 CFU·g⁻¹ dry soil in the agricultural soils. The

average number of culturable bacteria in the reference soil was 275 x 10⁵ CFU·g⁻¹ dry soil. The average count was significantly lower in agricultural soil than from reference soil. The number of culturable fungal populations was lower than the bacterial population and it varied between 67 x 10³ CFU·g⁻¹ dry soil to 260 x 10³ CFU·g⁻¹ dry soil in the agricultural sites (Table 2). The highest number of fungi found in the reference soil (284 x 10³ CFU·g⁻¹ dry soil). No specific pattern of the effect of brick kiln emission was observed against soil bacteria and fungi but an average decrease in total culturable numbers for the heavy metal contaminated soil samples indicate that Cd and Pb inhibit soil microbial population (Chen et al., 2014, Abdu et al., 2017). Bacteria seem to be more sensitive to heavy metal contamination than fungi (soil Cd and bacteria, r = -0.69, p < 0.001; soil Cd and fungi, r = -0.63, p < 0.001) (soil Pb and bacteria, r = -0.66, p < 0.001; soil Pb and fungi, r = -0.53, p < 0.001) (Table 3).

Total MA and SIR of soils were measured as microbial respiration rates (Table 2). The highest SIR was in site S18 (54 mg $CO_2 \cdot g^{-1}$) and the lowest in sites S1 (7.38 mg $CO_2 \cdot g^{-1}$). The average SIR in reference soil was 83.12 mg $CO_2 \cdot g^{-1}$. The brick kilns contribute significantly to the reduction of MA through its operation and emission (Khan et al., 2007). In this study, MA was not found to be significantly affected by brick kiln operation. The highest MA was in site S12 (28.18 mg $C \cdot g^{-1}$ soil·24h⁻¹) and the lowest in sites S1 (12.69 mg $C \cdot g^{-1}$ soil·24h⁻¹). The average MA in reference

Table	2
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Microbiological properties (mean ± SD) of different agricultural soils adjacent to the brick kiln cluster, Hathazari, Chattogram

Each mean is the average of the values obtained for three samples of each soil. Values in the same column followed by the same letter(s) are not significantly different at p < 0.05 according to ANOVA.

[CFU = Colony forming unit, SD = Standard deviation]

Sample	Number of culturable Bacteria	Number of culturable Fungi	Substrate induced respiration	Microbial activity	Microbial biomass carbon
	×10⁵ CFU·g⁻¹ dry soil	×10³ CFU·g⁻¹ dry soil	$mg CO_2 \cdot g^{-1}$	mg CO ₂ -C·g ⁻¹ day ⁻¹	mg C·kg⁻¹
S1	100.00 ± 1.00^{b}	86.30 ± 0.61^{b}	7.38±0.43 ^a	12.69±0.63ª	130.00±0.00ª
S2	$80.00{\pm}1.00^{\rm ab}$	$196.67 \pm 2.89^{\text{gh}}$	$13.05{\pm}0.71^{\rm cdef}$	21.45 ± 0.56^{h}	745.33±47.44 ^{ef}
S3	60.67±0.58ª	66.67±12.58ª	17.20 ± 1.11^{h}	13.00±0.51ª	338.70±44.47 ^{al}
S4	110.67±1.15°	248.33 ± 38.19^{ij}	10.00 ± 2.00^{ab}	$19.46{\pm}0.41^{\rm fg}$	687.33±58.35d
S5	$155.00 \pm 10.00^{\text{def}}$	260.00 ± 10.00^{jk}	10.00 ± 2.00^{ab}	17.09±0.55 ^{cd}	1500.00±300.0
S6	150.00 ± 10.00^{de}	210.00 ± 10.00^{h}	9.33±1.15 ^{ab}	13.48±0.53ª	300.00±100.00
S7	138.33 ± 7.64^{d}	170.00 ± 5.00^{de}	11.33 ± 1.15^{bcd}	$19.13{\pm}0.35^{\rm efg}$	600.00±200.00
S8	301.67 ± 12.58^{j}	251.33±2.08 ^{ij}	8.00±0.00 ^a	20.10 ± 1.19^{g}	1600.00±200.0
S9	231.67 ± 7.64^{i}	178.33 ± 2.89^{def}	10.00 ± 2.00^{ab}	22.91 ± 0.63^{i}	1666.67±115.4
S10	$173.33 \pm 7.64^{\rm fg}$	166.67 ± 7.64^{de}	12.67±1.15 ^{cde}	15.32±0.65b	566.67±208.17
S11	200.00 ± 20.00^{h}	183.34 ± 0.15^{efg}	13.33 ± 1.15^{def}	26.63±1.37 ^j	900.00±100.00
S12	163.33±25.17 ^{efg}	160.00 ± 5.00^{d}	$14.00{\pm}2.00^{\rm efg}$	28.18 ± 0.83^{k}	1600.00±200.0
S13	105.00±13.23°	203.00 ± 2.65^{h}	$15.33 \pm 1.15^{\text{fgh}}$	$19.04{\pm}0.58^{\rm efg}$	679.67±57.77 ^d
S14	80.00 ± 10.00^{ab}	106.67±0.58°	$14.00{\pm}2.00^{\rm efg}$	15.41±0.61 ^b	466.67±152.75
S15	206.67 ± 30.55^{h}	$194.00{\pm}3.61^{\rm fgh}$	$16.00 \pm 2.00^{\text{gh}}$	23.45 ± 0.31^{i}	900.00±300.00
S16	106.67±1.53°	81.00 ± 9.54^{ab}	17.33 ± 1.15^{h}	15.27 ± 1.67^{b}	300.00±100.00
S17	113.00±1.00 ^c	165.00 ± 5.00^{de}	$10.67 \pm 2.31^{\rm hc}$	16.55 ± 0.55^{bc}	950.00 ± 20.00^{f}
S18	$185.00 \pm 22.91^{\text{gh}}$	240.00 ± 10.00^{i}	54.00 ± 0.00^{i}	$18.26{\pm}0.07^{\rm def}$	806.67±66.58e
S19	281.33±0.58 ^j	295.33±1.53 ^m	76.40 ± 0.40^{j}	18.00 ± 1.14^{de}	1765.67±152.1
S20	252.67 ± 4.16^{i}	272.00 ± 2.00^{kl}	87.27 ± 1.10^{k}	23.09 ± 0.00^{i}	2314.37±100.1
S21	290.00+10.00 ^j	283.33+2.89lm	85 69+1 48 ^k	22 55+0 00 ^{hi}	2513 33+66 58

Table 3

Correlation coefficients among soil physicochemical characteristics, Pb and Cd concentrations and soil microbial and enzymatic properties

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Soil	Soil microbial and	Correlation equation	\mathbb{R}^2	r
properties	enzymatic properties			
pH	Bacteria	y = 66.75x – 180.85	0.31	0.55ª
	Fungi	y = 37.06x – 1.28	0.11	0.34^{b}
	SIR	y = 13.70x – 46.77	0.10	0.32 ^b
	MA	y = 2.66x + 5.29	0.14	0.37^{b}
	MBC	y = 811.21x - 3200	0.52	0.72 ^a
	Dehydrogenase	y = 217.39x – 681.88	0.48	0.69 ^a
	Urease	y = 10.48x – 19.13	0.14	0.38 ^b
	AcidP	y = 77.00x – 244.68	0.34	0.58ª
	ArylS	y = 14.42.42x – 25.89	0.43	0.66ª
OM	Bacteria	y = 131.57x + 29.75	0.27	0.52ª
	Fungi	y = 123.23x + 63.76	0.28	0.53ª
	SIR	y = 10.57x + 13.49	0.01	0.12 ^{ns}
	MA	y = 9.35x + 9.42	0.39	0.62ª
	MBC	y = 1623.6x – 665.03	0.47	0.68 ^a
	Dehydrogenase	y = 449.72x – 17.69	0.46	0.68^{a}
	Urease	y = 31.37x + 2.87	0.29	0.54ª
	AcidP	y = 101.03x + 50.88	0.13	0.36^{b}
	ArylS	y = 27.54x + 20.54	0.35	0.60ª
Clay	Bacteria	y = 7.35x – 26.56	0.20	0.44ª
	Fungi	y = 5.01x + 59.95	0.11	0.33^{b}
	SIR	y = 0.91x + 0.51	0.02	0.15 ^{ns}
	MA	y = 0.31x + 10.90	0.10	0.32 ^b
	MBC	y = 76.86x – 998.10	0.25	0.50ª
	Dehydrogenase	y = 17.80x – 18.64	0.17	0.41 ^a
	Urease	y = 0.76x + 15.66	0.04	0.20 ^{ns}
	AcidP	y = 2.72x + 84.32	0.02	0.15 ^{ns}
	ArylS	y = 1.50x + 9.62	0.25	0.50 ^a
Soil Pb	Bacteria	y = -4.27x + 280.57	0.44	-0.66ª
	Fungi	y = -3.10x + 274.58	0.28	-0.53ª
	SIR	y = -1.58x + 66.86	0.47	-0.68ª
	MA	y = -0.22x + 25.09	0.34	-0.58ª
	MBC	y = -49.55x + 2345.90	0.67	- 0.8 2ª
	Dehydrogenase	y = -11.41x + 754.10	0.46	-0.68 ª
	Urease	v = -0.75x + 55.52	0.26	-0.51ª
	AcidP	v = -4.10x + 265.55	0.33	-0.58ª
	ArylS	y = -0.63x + 66.04	0.29	-0.54ª
Soil Cd	Bacteria	y = -168.90x + 261.74	0.48	-0.69ª
	Fungi	y = -138.75x + 270.02	0.39	-0.63ª
	SIR	y = -48.96x + 52.20	0.32	-0.56ª
	MA	y = -11.03x + 25.36	0.59	-0.77ª
	MBC	v = -1910.10x + 2099.00	0.71	-0.84ª
	Dehydrogenase	y = -543.88x + 756.32	0.74	-0.86ª
	Urease	y = -33.069x + 54.11	0.35	-0.60ª
	AcidP	y = -133.57x + 231.22	0.25	-0.50ª
	ArvlS	y = -27.42x + 64.60	0.38	_0.50
PER	Bacteria	v = -0.75x + 265.74	0.49	0 71a
	Fungi	y = -0.61x + 272.38	0.30	_0.71
	SIR	y = -0.22x + 54.14	0.35	_0 5&a
	MA	y = -0.22x + 34.14 y = -0.05x + 25.40	0.54	-0.36*
	MBC	y = -0.03 + 23.43 y = -8.53 + 21.47.10	0.57	_0.70"
	Debudrogonaco	$y = -0.33x \pm 214/.10$ $y = -2.38y \pm 762.96$	0.72	-0.00"
	Denyurogendse	y = =2.30X + 703.00	0.72	-0.00"
	Uronec	x = 0.15x + 54.70	0.25	0 500
	Urease	y = -0.15x + 54.70	0.35	-0.59ª

^a Correlation is significant at the 0.001 level ^b Correlation is significant at the 0.01 level ^c Correlation is significant at the 0.05 level

[*PER* = comprehensive potential ecological risk index, SIR = Substrate induced respiration, MA = Microbial activity, MBC = Microbial biomass carbon, AcidP = Acid phosphatase, ArylS = Arylsulfatase] soil was 21.21 mg C·g⁻¹ soil·24h⁻¹. There was no significant difference between reference and agricultural soils for average MA content. The activity and biomass of soil microbes are closely related to soil fertility and environmental quality. There was a significant difference between agricultural soils near brick kiln and reference soil with the MBC. The amount of MBC in the agricultural soil samples near brick kiln ranged from 130.00 to 1666.67 mg biomass C·kg soil⁻¹, which were lower values than those of the average values of reference soils (2197.79 mg biomass C·kg soil⁻¹).

Several studies reported a decrease in the number of culturable bacteria and fungi with increasing levels of heavy metal contamination (Yuan et al., 2015; Srivastava et al., 2017). According to Ali et al., (2019), the way heavy metals act depends on their type and rate. Pb doses above 50 mg·kg⁻¹ decreased the count of both bacteria and fungi. Khan et al., (2008) showed the inhibitory effect of high Cd and Pb concentrations on soil MBC. Heavy metals decrease MBC and reduce their activity in the soil (Wyszkowska et al., 2008). Soil microbial biomass carbon is a sensitive parameter and can be used as an indicator of changes in OM composition (Yang et al., 2006). The decrease in MBC caused by a high level of heavy metal contamination found at the sites agrees with Wang et al., (2007) and Srivastava et al., (2017). The synthesis of MBC in soils polluted by heavy metals can be less effective than in non-polluted soils due to the stress caused by heavy metals. Yang et al., (2006) stated that Cd and Pb cause disorders in the soil MA and depress the MBC of microorganisms.

Microbial indices are useful quantitative tool to assess the effects of anthropogenic stress or disturbance on the soil microbial community. The qMic values decreased, expressing the maintenance energy, as the amount of heavy metal in soil increased. The qMic value was the highest in site S8 (2.26%) and the lowest in S1 (0.34%). In reference soil, the average amount

was 3.14%. To survive in stress, soil microorganisms reduce the conversion of substrate into new MBC and other metabolic processes, therefore qMic decreased (Fig. 3). A reduction of this ratio as a result of metal pollution has been reported from other studies (Anderson and Domsch, 1989; Brookes, 1995). Soil microorganisms develop protective measures to adapt in long-term heavy metal pollution by physical exclusion by exopolymers, intracellular sequestration with low molecular weight compounds and precipitation of metals as phosphates, carbonates, and sulfides (Yang et al., 2006; Wang et al., 2007). This kind of cellular activity requires huge energy that increases the demand for maintenance energy. Therefore, qMic also shows the survival capacity of soil microorganisms. Soil pollution due to heavy metal contamination is a serious problem because tolerant microorganisms can bioaccumulate heavy metals that directly affect the food chain to human health (Zhuang et al., 2009). On the other hand, soil microorganisms under environmental stress shift more energies from growth and reproduction to maintenance and survival, leading to an increase of qCO_2 (Zhao et al., 2020). Our results show that the qCO₂ increased markedly with increasing heavy metal concentration (Fig. 3). qCO, was the highest in S1 (9.76 mg CO₂–C·mg⁻¹ Cmic·h⁻¹ × 10⁻⁴). In reference soil, average qCO₂ was 0.97 mg CO₂-C·mg⁻¹ Cmic·h⁻¹ × 10⁻⁴. According to the finding, qCO_2 was generally lower in reference soil compared to the average agricultural soils (3.22 mg CO₂–C· mg⁻¹ Cmic·h⁻¹ × 10⁻⁴) near the brick kiln cluster. Reference soils were free from the negative impact of anthropogenic activities during the study period. Liao and Xie (2007) and Zhang et al., (2008) also found the qCO_2 as a good indicator of the negative impact of the heavy metal pollution on the soil microorganisms. Correlation study also demonstrated that qCO_2 was negatively correlated with soil MBC, number and activity but qCO_2 was significantly positively correlated with heavy metals (soil Cd and soil Pb) (soil Cd and *q*CO₂, r = 0.68, p < 0.001; soil Pb and qCO₂, r = 0.68, p < 0.001).



Fig. 3. Relationship of microbial biomass carbon (MBC) and microbial indices (*q*MIC and *q*CO₂) in agricultural soils beside brick kiln cluster, Hathazari, Chattogram. Regression equation, line of the best fit and R² is shown

3.4. Enzymatic properties of the agricultural soils

Soil enzymes are biological active soil components that have an intimate association with physicochemical and biological soil characteristics (Shukla and Varma, 2011). Soil enzyme activity is widely used as a biological indicator to assess soil contamination. Generally, high enzyme activity represents good soil quality (Fazekašová and Fazekaš, 2020). The dehydrogenase, urease, acid phosphatase and arylsulfatase, enzymes involved in the C-N-P-S cycle in soil varied widely among the soils studied. The results showed that the average enzymatic activities in agricultural soils were lower than the reference sites. The level of enzyme activity varied in a wide range and for dehydrogenase amounted 94.00 to 718.00 mg formazan·kg soil⁻¹24h⁻¹, for urease 14.29 to 65.71 mg NH₄–N·kg soil· 2h⁻¹, for acid phosphatase 56.27 to 318.57 mg p- nitrophenol kg soil⁻¹ h⁻¹, and for arylsulfatase 25.64 to 70.87 mg p- nitrophenol kg soil⁻¹ h⁻¹ (Table 4). On the reference sites mean concentration of dehydrogenase, urease, acid phosphatase and arylsulfatase was 616.22 mg formazan·kg soil-124h-1, 47.38 mg NH₄–N·kg soil·2h-1, 264.22 mg p- nitrophenol kg soil ⁻¹ h⁻¹, and 64.68 mg p- nitrophenol kg soil ⁻¹ h⁻¹ respectively. A significant positive correlation (p < 0.001) between dehydrogenase, urease, acid phosphatase and arylsulfatase was found in this study (Fig. 2b). Agricultural soils with high heavy metal concentration showed reduced soil enzyme activities. Cd and Pb had shown a very high significant negative correlation with the enzymes- dehydrogenase, urease, acid phosphatase and arylsul-

fatase (Table 3). Soil enzyme activities are sensitive to the high level of heavy metals (Wang et al., 2008; Xian et al., 2015; Srivastava et al., 2017). The application of different rates of Cd, Pb, and Cd/Pb mixture in soil reduced the activities of acid phosphatase, urease and MBC in comparison to reference soil (Pan and Yu, 2011). All enzyme activities can occur extracellularly along with or within a living cell whereas dehydrogenase activity only acts inside a living cell (Garcýa-Gil et al., 2000; Wang et al., 2007). Therefore, microbial activity inhibited by heavy metal stresses directly express the less dehydrogenase activity.

ED₅₀ value is used to measure the sensitivity of an ecosystem to stress. When a basic ecological function is reduced to 50% for external stress, the stress is labelled as extreme for its continued functioning. Table 5 shows the ED₅₀ values for the four enzyme activities as measured for the heavy metal contaminated agricultural soils and the best fit model for ED₅₀ values was selected from higher R² values from the regression analysis. Studies on the impact of toxic metals on soil enzyme showed that enzyme activities in a stressed ecosystem would be always less than 100% of the reference value. Model 1 was a full inhibition model. Model 2 was a partial inhibition model, suggesting that a fraction of the enzymatic activities was not inhibited by heavy metal contamination to the soil. Model 3 indicates that the relationship between enzyme activity and the toxicity coefficient is sigmoidal dose. It is very hard to interpret the exact reason for the decrease in soil enzyme activities as it may be due to a direct metal inhibition to enzymes or a lower synthesis and re-

Table 4

Soil enzymes involved in soil C (dehydrogenase), N (urease), P (acid phosphatase) and S (arylsulfatase) (mean ±SD) turnover in soils in different agricultural soils adjacent to the brick kiln cluster, Hathazari, Chattogram

Each mean is the average of the values obtained for three samples of each soil. Values in the same column followed by the same letter(s) are not significantly different at p<0.05 according to ANOVA.

Sample	Dehydrogenase	Urease	Acid Phosphatase	Arylsulfatase
	mg formazan·kg soil-1 24 h ⁻¹	mg NH₄–N·kg soil∙ 2h⁻¹	mg p- nitrophenol kg soil ⁻¹ h ⁻¹	mg p- nitrophenol kg soil ⁻¹ h ⁻¹
S1	140.33±4.73ª	15.24±0.82ª	81.64±0.30 ^a	52.52 ± 6.83^{ef}
S2	566.00±22.61 ⁱ	35.00±0.71°	123.72±5.77 ^{ab}	58.63 ± 9.95 fg
S3	139.00±43.14ª	15.00±0.71ª	147.71 ± 2.88^{ab}	$41.88{\pm}4.04{}^{\rm cd}$
S4	$425.67 \pm 18.61^{\rm f}$	35.00±0.71°	80.04 ± 1.08^{a}	41.45 ± 3.13 ^{cd}
S5	$465.00 \pm 15.10^{\text{gh}}$	29.05 ± 0.82^{d}	207.76 ± 13.95^{bcde}	$59.95 \pm 1.31^{\text{gh}}$
S6	254.67±13.58°	20.95 ± 0.82^{b}	93.41±1.12ª	34.23±0.53 ^b
S7	320.67 ± 0.58^{d}	15.00±0.71ª	7695.00±21.54ª	51.49±1.75 °
S8	697.33±2.52 ^k	15.00±0.71ª	$306.56 \pm 11.01^{\rm f}$	62.47 ± 4.02 gh
S9	684.67 ± 1.53^{k}	65.00±0.71 ^j	175.26 ± 7.34^{abcd}	63.53 ± 0.05 gh
S10	359.33±4.16 ^e	30.00 ± 1.43^{d}	100.80 ± 19.47^{a}	37.88 ± 4.02 bc
S11	568.67 ± 23.12^{i}	55.00 ± 0.71^{h}	97.33±21.73 ^a	33.27±1.49 ^b
S12	634.67 ± 9.02^{j}	50.00±1.43 ^g	124.33±1.87 ^{ab}	47.25 ± 2.83^{de}
S13	333.33±0.58 ^{de}	50.00±1.43 ^g	135.51±28.20 ^{ab}	$33.09 \pm 5.53^{\mathrm{b}}$
S14	153.67±9.29ª	$40.00 \pm 1.43^{\rm f}$	118.75 ± 5.51^{ab}	25.70±0.07 ª
S15	478.67 ± 1.15^{h}	50.00±1.43 ^g	145.26±11.31 ^{ab}	66.49 ± 3.95 h
S16	215.33±11.59 ^b	30.00 ± 1.43^{d}	142.68 ± 27.21^{ab}	36.14 ± 1.55 bc
S17	677.00 ± 37.32^{k}	25.00±0.71°	159.32 ± 15.29^{abc}	42.35 ± 0.77 cd
S18	$442.67 \pm 16.01^{\rm fg}$	25.00±0.71°	155.30±2.17 ^{abc}	$47.62 \pm 4.45^{\text{ de}}$
S19	567.33 ± 39.11^{i}	20.00 ± 1.43^{b}	277.44±8.65 ^{ef}	60.80 ± 0.14 gh
S20	602.33±33.86 ^j	63.57±0.71 ^j	$266.08 \pm 230.565^{\rm def}$	$66.60 \pm 0.37^{\mathrm{h}}$
S21	679.00 ± 9.85^{k}	58.57 ± 1.43^{i}	249.16 ± 11.64^{cdef}	66.63 ± 1.96 h

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Table 5

Values of R^2 (p < 0.05) obtained for Gauss-Newton analysis, which best describes the inhibition of dehydrogenase, urease, acid phosphatase, and arylsulfatase of the agricultural soils in the vicinity of brick kiln cluster and ED₅₀ (mg·kg⁻¹) expressed by total ecological toxicity coefficient, *PER*

Brick kiln operation impact on soil microbial biomass and enzyme activity

Soil Enzymes	Model	ED ₅₀	R.
Dehydrogenase	1	299.75	0.69
Urease	2	633.28	0.72
	3	421.69	0.58
	1	32.21	0.35
	2	43.85	0.36
	3	19.94	0.31
Acid phosphatase	1	306.39	0.26
	2	507.81	0.30
	3	376.44	0.30
Arylsulfatase	1	124.85	0.38
	2	104.69	0.40
	3	93.15	0.36

lease of enzymes, or a combination of both (Gao et al., 2010). Soil enzymes can be physically and chemically blocked by active organic and inorganic ligands (Renella et al., 2003). The ED_{so} values for dehydrogenase, urease, acid phosphatase and arylsulfatase activity were predicted with Model 2. The dehydrogenase was most sensitive to metal addition and easily lost activity under low heavy metal concentration, urease and arylsulfatase activity were also sensitive to the combined heavy metal effect. The effect of heavy metal on acid phosphatase was found to be lower than the other enzymes as phosphatase activity was high heavy metal rate responsive to inhibit it. A significant relationship between soil enzymes and PER as fitted by Model 2, also indicates an adaptation of soil microorganisms in our study area. Sensitivity to metal toxicity varies for soil microorganisms. Sensitive populations decrease and remunerate with the increase of metal-tolerant strains in heavy metal stressed soil system (El Baz et al., 2015). The results of many experimental studies suggest that inhibition of soil enzymes due to heavy metal contamination can be reduced over time, and some microorganisms could be adapted to long term polluted environments and thereby help enzymatic activity to recover (Fazekašová and Fazekaš, 2020). The microorganism in our study site might be adapted to the high heavy metal concentration. Brick production activities are a long time regular practice in this zone. Cd concentration was found to be a very high toxic level in some highly contaminated soils enough to suppress complete microbial activity, but due to their adaption capacity, they survived with limited enzyme activity.

3.5. Effects of soil properties and heavy metal contamination on size and activity of microbial biomass and enzyme activity

Soil microorganisms are exposed to heavy metals via the soil solution with the ultimate impact on soil microbial biomass and soil enzyme activity. The concentration of heavy metals in soil solution depends upon soil processes mediated by OM, pH and nutrient availability. Therefore, the type of soils plays a major role in the toxic effect of heavy metals on soil microorganisms. Some significant relationships between soil OM, MBC and MA were found in the studied soils (Table 3) (p < 0.001), which are involved in soil metabolic processes. Microbial population and activity and enzyme activities were correlated positively with soil pH (Table 3). Total nitrogen and AvP had a positive correlation with the number of culturable bacteria and fungi and SIR. Similar results have been obtained by Taylor et al., (2002). Soil pH can affect the adsorption/desorption of soil ions and soil nutrient transformation (Li et al., 2018). Chen et al., (2014) and Xian et al., (2015) showed the positive relationship between OM and dehydrogenase and arylsulfatase enzyme activity.

All the four enzymes were positively correlated (p < 0.05) with bacterial and fungal populations, SIR, MA, MBC (Fig. 2b). Dehydrogenase activity was significantly high in reference soil, where the soil microbiota was also metabolically more active than in contaminated agricultural soils (Table 2, 4). Since dehydrogenase is an intracellular enzyme involved in microbial metabolism, its lower activity in agricultural soil may be related to the smaller MBC content, but also a larger heavy metal concentration in agricultural than in reference soils. Furthermore, the dehydrogenase activity was significantly correlated with soil microbial biomass C (r = 0.61, p < 0.05). The decrease of soil MBC and inhibition of dehydrogenase activity have been reported also in polluted areas near an aluminium smelter (Tscherko and Kandeler, 1997) These results suggest that MBC and dehydrogenase can be useful measures of the level of heavy metal contamination in a soil sample. Arylsulfatase, urease and acid phosphatase activity also showed a positive correlation with MBC (p < 0.05), MA and qCO_2 (p < 0.05).

The results of this study found suppression of all the soil microbial properties and enzyme activities, which indicated the disruption of soil function by the Cd and Pb contamination in the vicinity of brick kiln cluster sites. The qCO_2 , which expresses the stress situation on soil microorganisms, with increasing *PER* was fitted by the exponential curve (Fig. 4). Regression analysis produced significant relationships between *PER* versus all the microbial properties and enzyme activities (Table 3). The microbial population as reduced by heavy metals which in turn showed a decrease in the activities of soil enzymes. Consequently, the decomposition rate of carbon, nitrogen, phosphorus and sulfur in soils would be blocked. Different enzymes have differ-

Fig. 4. Relationship of qCO_2 (metabolic quotient) with *PER* (comprehensive potential ecological index) in different agricultural soils besides brick kiln cluster, Hathazari, Chattogram. Regression equation, line of the best fit and R² is shown. Filled circles representing the average qCO_2 values and open circles representing the replications



ent characteristics and mechanisms. The effects of heavy metal contamination on enzyme activities can be mediated by soil pH (Dick, 2011), OM content (Tang et al., 2020) and clay content (Tietjen and Wetzel, 2003). Soil enzyme activity is an effective indicator of soil quality resulting from environmental stress or management practices (Karaca et al., 2010). Thus, soil pH is one of the very important factors that are considered to evaluate the effect of pollutants on the activity of soil microorganisms. However, it is very difficult to separate the effect of heavy metal stress on soil microbial populations from that due to pH changes (García Gil et al., 2013). Acidic pH levels may intensify the heavy metal effect further (Wyszkowska et al., 2016). The low pH leads to the increased bioavailability of Cd and Pb in soil (Humberto et al., 2020), which results in higher heavy metal toxicity for microorganisms and inhibition of enzyme activities. The nutrient content of the soil has a regulatory effect on the toxicity of heavy metals in soil (Chodak et al., 2013). But in our findings (Table 3), the effect of OM, TN, AvP on microorganisms was not as strong as that of Cd and Pb content. Besides, the non-uniform distribution of brick kilns activities related to land-use practices causes some anomalous soil properties. The concentration of the Cd and Pb was not found in very high contamination levels as reported by Ismail et al., (2012) and Ravankhah et al., (2017) which predicts the presence of other contaminants. Detailed studies on other harmful heavy metals as well as the evolution of polycyclic aromatic hydrocarbons (PAHs) from brick kiln operation should be considered for future studies.

3.6. Hierarchical cluster analysis (HCA) and Principal component analysis (PCA) of the study sites

The entire study was located in an agricultural area without significant differences in the type of soil texture which expresses the origin of the soils from the similar parent material. The concentrations of Cd and Pb were greater than background concentrations indicating their distribution was most likely directly related to anthropogenic source (Ravankhah et al., 2017). Dendrogram grouping of heavy metal contaminated soils characterized by similar responses of soil microorganisms and their enzyme activity to heavy metal concentration and respective heavy metal indices along with soil physicochemical properties were performed (Fig. 5). The dendrogram revealed two main clusters of similarities with heavy metal contaminated soils. A cluster that contained S19, S20 and S21 along with S8 was significantly different from the other agricultural soils. The sites that are clustered together with the reference sites, we can tell that the environmental situation in these soils with a combination of Cd and Pb concentration with physicochemical characteristics of the soils, were not under any stressed condition.

Principal component analysis has commonly been used for investigating metal sources, anthropogenic activities, or soil parent materials (Kormoker et al., 2019). The depositions of atmospheric particulates released by brick kiln clusters were believed to contribute to Cd and Pb in investigated agricultural soils. A PCA was performed on a correlation matrix of the data obtained on soil microbial and enzymatic activities affected by soil physicochemical properties and soil Cd and Pb content (Fig. 6). The PCA revealed a strong relationship between enzymatic activities and soil microbial properties, and they varied together in the same trend upon toxic impact from heavy metals. The PCA analysis showed that microbial biomass and activities and all the enzyme activities were highly associated with soil pH, OM, and TN. Cd and Pb were significantly and positively associated with the heavy metal indices and qCO_2 . Dehydrogenase, acid phosphatase and arylsulfatase responded similarly to soil contamination with heavy metals, which is demonstrated by the proximity of vectors representing the analyzed enzymes. Urease and MA were also sensitive to heavy metals, but its response to heavy metals was somewhat different. This is illustrated by the position of the MA and urease vector relative to the soil heavy metal indices (PER). Distribution of sampling area in the PCA plot manifests the difference between the sampling sites concerning pollution levels and the impact on soil microbial biomass and enzyme activity. Gao et al., (2010) showed that absolute enzymatic activities varied under different land uses depending on the types of land use or management and the type of enzyme.



Fig. 5. Dendrogram of the hierarchical cluster analysis of Cd and Pb concentration in sampling sites with their impact on soil microbiological properties



Fig. 6. PCA plot showing the similarity of agricultural soils around brick kiln cluster considering Cd and Pb concentration in sampling sites with their impact on soil microbiological properties

4. Conclusions

The brick kiln industry has paramount importance in Bangladesh for modern establishments due to rapid urbanization and residential construction. The present study provided a comprehensive insight into the responses of soil microbial biomass and soil enzyme activities in agricultural soils related to brick kiln operation. This study showed that:

- Agricultural soil quality was low in the vicinity of brick kiln cluster than the reference sites concerning for soil OM and essential nutrients concentration
- The study area was contaminated with Cd and Pb (up to 1.07 mg·kg⁻¹ soil and 52.07 mg·kg⁻¹ soil respectively) as compared to their background levels
- The heavy metal concentrations varied at the different agricultural sites but display a similar variation pattern
- Pb concentration was found within the acceptable limit in soils when compared to the Chinese environmental quality standards for soil. On the other hand, Cd concentration was very high in the sampling areas.
- The *PER* which shows the extent of contamination by the heavy metals indicated moderate to a considerable level of contamination in the agricultural soils around the brick kiln cluster
- The intensive uncontrolled operation of brick kiln in the study area is a potential source of pollution in terms of heavy metal contamination
- The number of bacteria and fungi, respective the intensity of enzymatic activities and the microbial biomass carbon decreased with increasing heavy metal concentrations
- Microbial indices (*q*MIC and *q*CO₂) suggests the heavy metal stress as well as some microbial adaptation in the study area
- Cd was the most effective inhibitor on the dehydrogenase, arylsulfatase and acid phosphatase enzymes in this study
- The most sensitive to pollution was found to be with soil bacteria, MBC and dehydrogenase activities

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